

# Ferrocenyl based Polyvinylbenzyl Chloride Nanofiber for Decyanidation

Adesoji, Adedoyin O.<sup>a,b,c\*</sup>, Amolegbe, Saliu A.<sup>b</sup>, Tshentu, Zenixole<sup>c</sup> and Adewuyi, Sheriff<sup>b</sup>

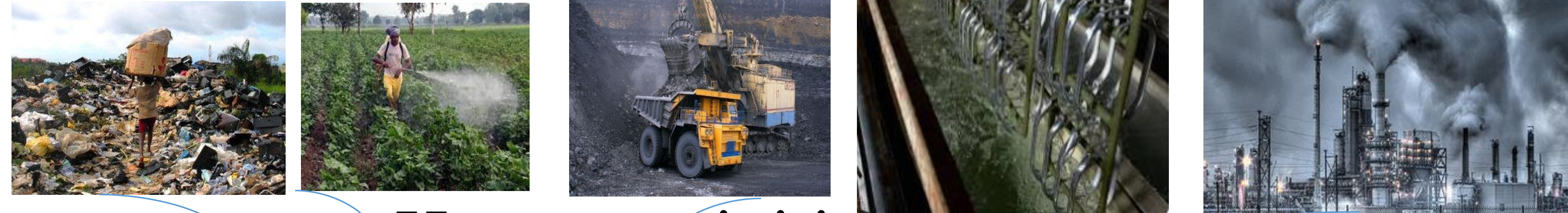
a. Chemistry Unit, Distance Learning Institute, University of Lagos, Lagos, Nigeria

b. Chemistry Department, Federal University of Agriculture, Abeokuta, Ogun State, Nigeria

c. Chemistry Department, Nelson Mandela University, Port Elizabeth, South Africa



## Introduction



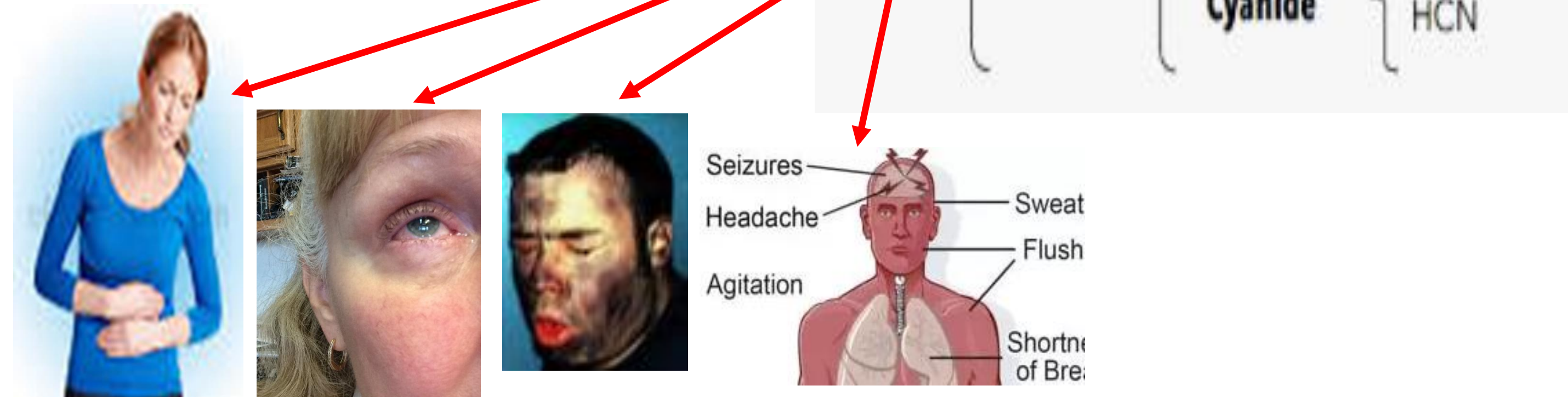
Human activities

ground, atmospheric and aquatic ecosystem

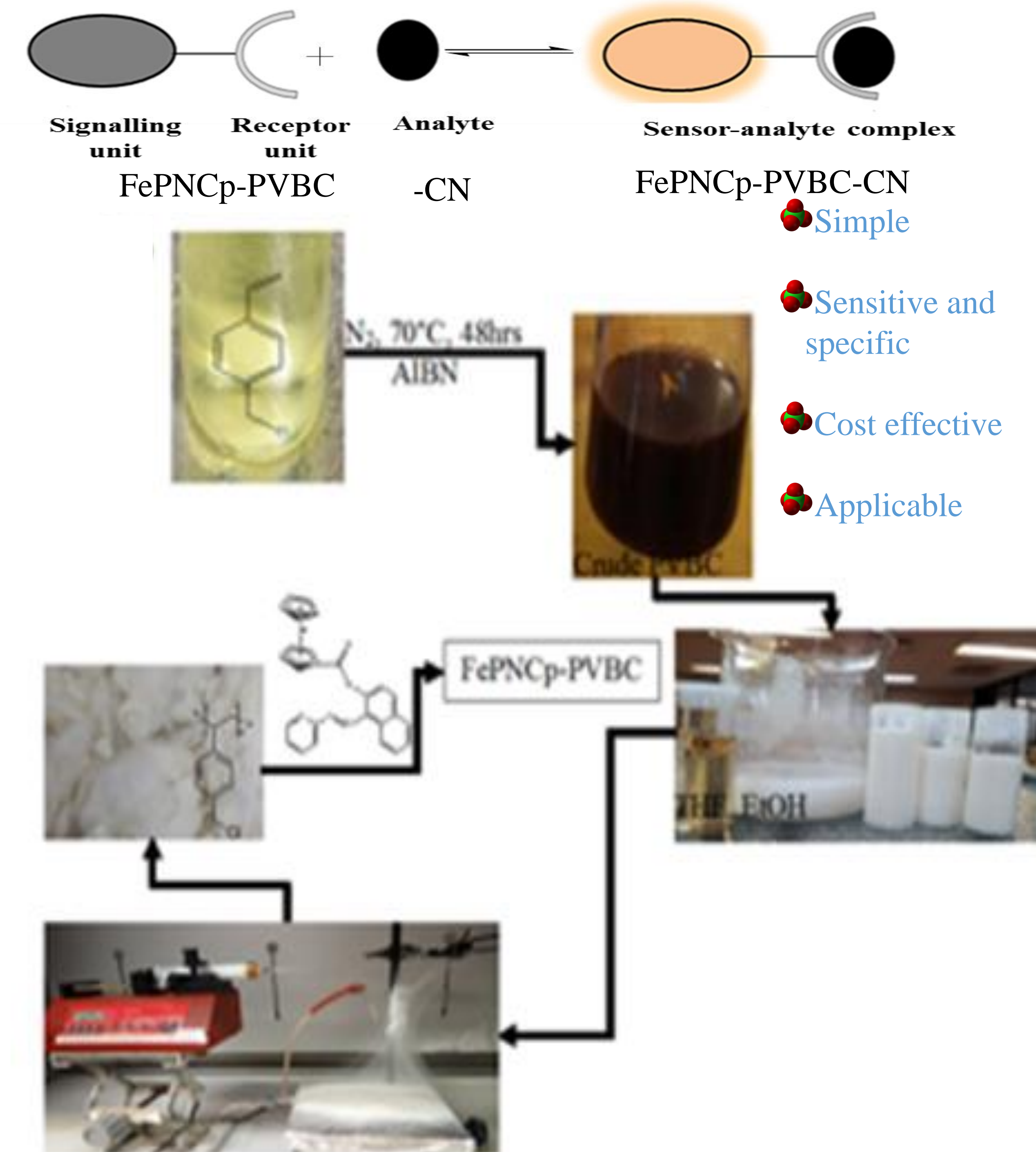
ecological and human-health threats

Contaminants (cations, anions)

Toxic CYANIDE



## Methodology



## Result and Discussion

The synthesis of EP was done via diazotization and esterification reaction. Immobilization of electrospun polyvinylbenzyl chloride (PVBC) on EP produced FePNCp-PVBC (PEP) which was further applied for cyanide removal. The ligand, EP and functionalized material, PEP were characterized using analytical techniques.

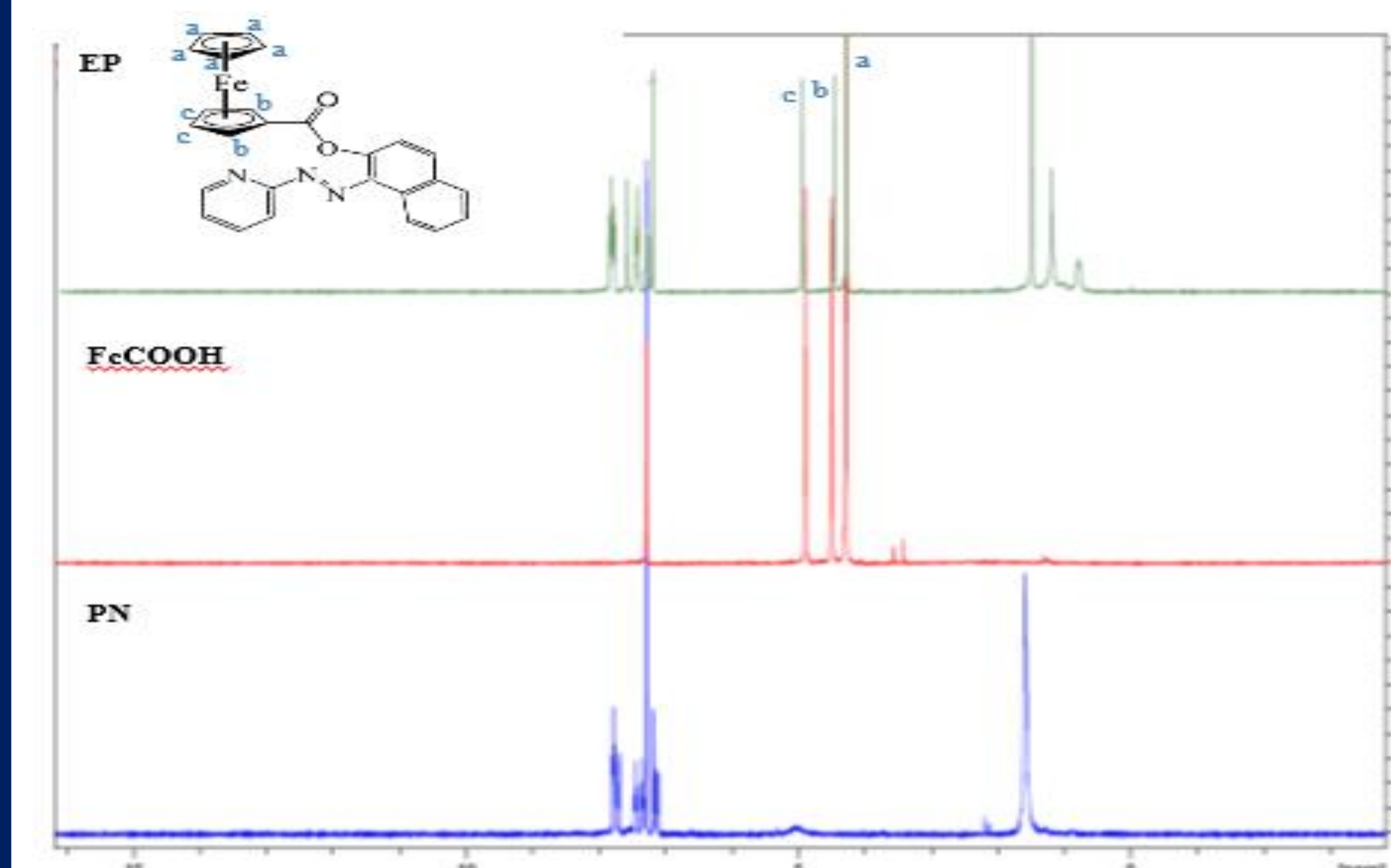


Fig. 1: Proton NMR Spectra of EP and reactants

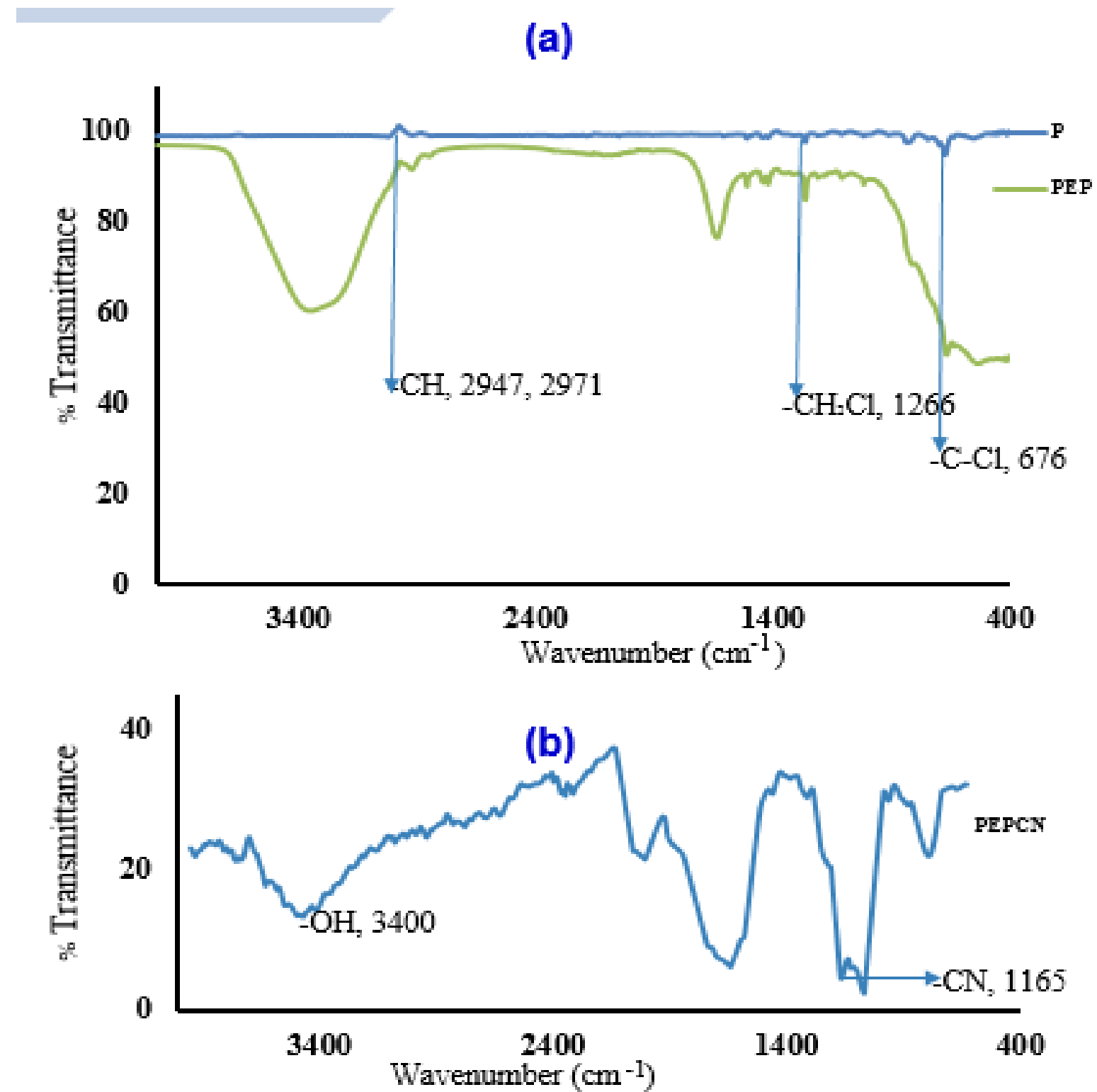


Fig.2: (a) FTIR Spectra of P & PEP (b) PEP

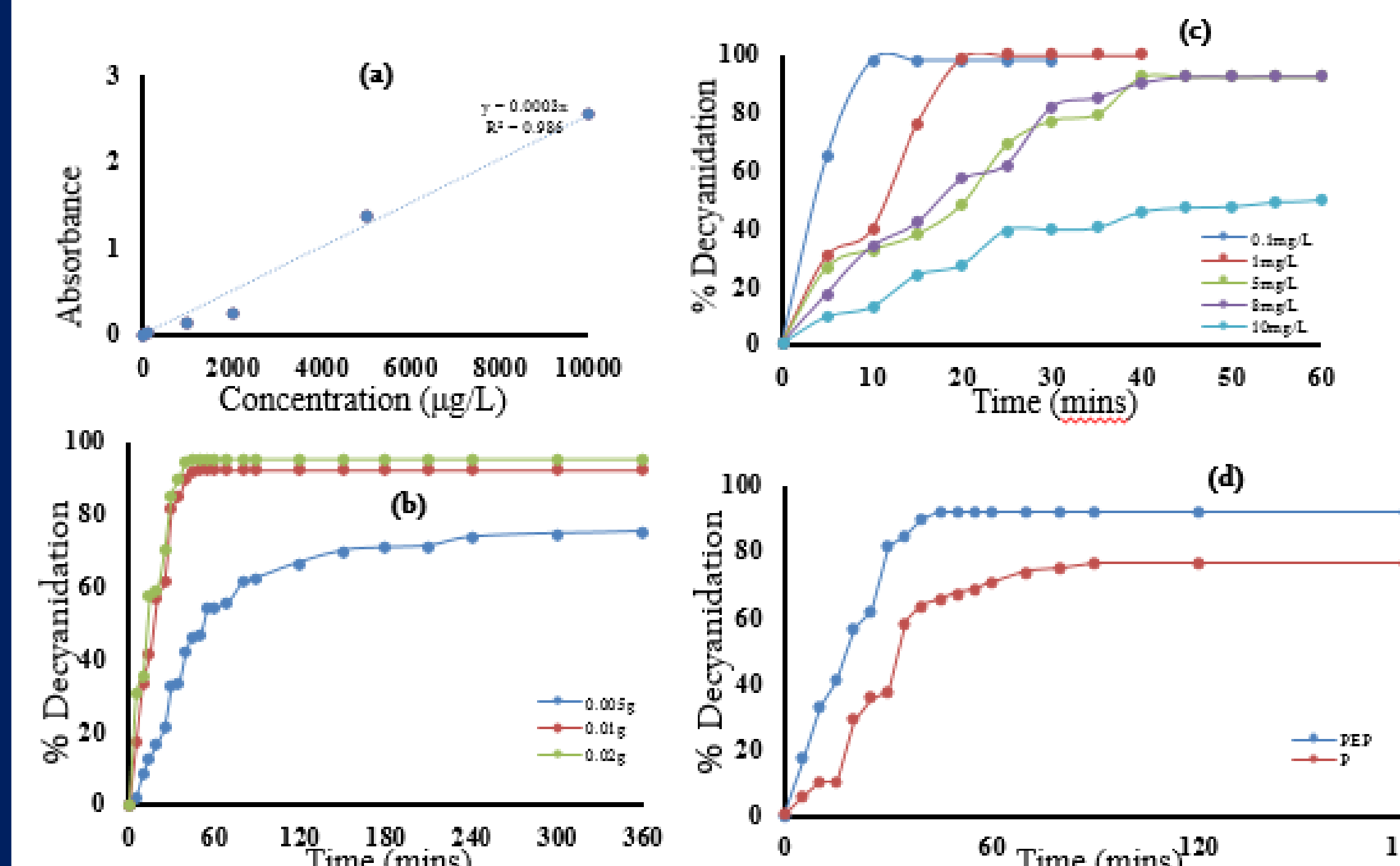


Fig. 3: (a). Calibration curve for varying cyanide concentrations. (b): Decyanidation capacity (DC) of PEP of different weights; 0.005g, 0.01g and 0.02g (room temperature, rotational speed of 250 rpm, initial cyanide concentration of 8mg/L) (c) DC of PEP at different initial concentrations. (d) DC of 0.01g P & PEP at 8mg/L CN.

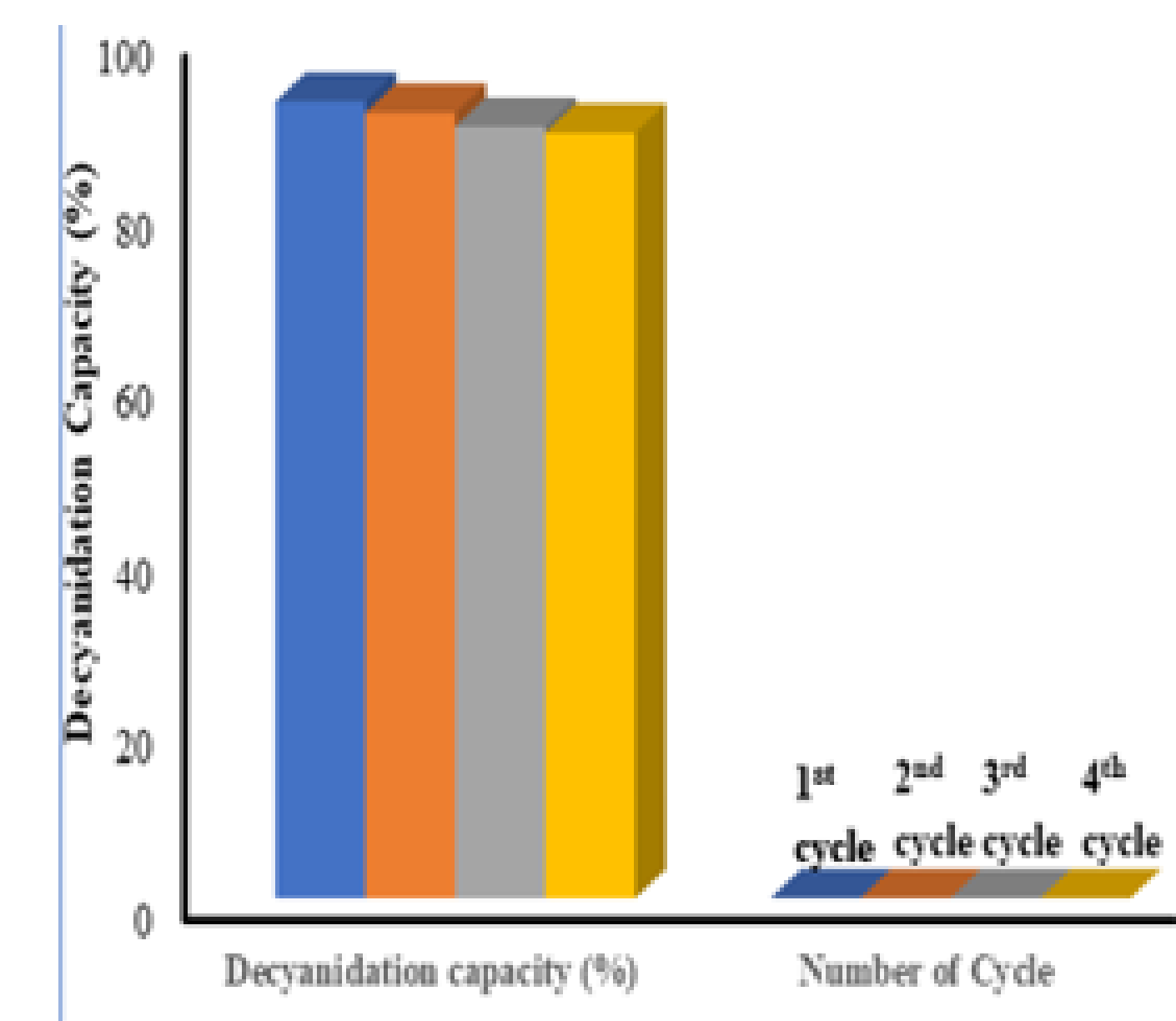


Fig. 4: The bar chart of the 0.01 g FePNCp-PVBC (PEP) showing the cyanide removal at an initial concentration of 8 mg/L CN for four cycles

## Conclusion

❖ Successful synthesis and characterization of an ester-based ferrocenyl ligand, FePNCp, and its subsequent immobilization on the electrospun polyvinylbenzyl chloride (PVBC) as a functionalized nanofibrous material, FePNCp-PVBC (PEP) was done.

❖ The response time of FePNCp-PVBC (PEP) proved effective in 45 minutes, even with 0.01g of the adsorbent dosage, and showed a 92.2 % decyanidation capacity.

## Selected References

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