## Graphitization as a way to stabilize textural characteristics of alumina under hydrothermal conditions

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 $\gamma$ -Al<sub>2</sub>O<sub>3</sub> is a widely used support for catalysts of various processes. The significant drawback of  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> is its instability in hydrothermal conditions, which might be essential for biomass refining processes. In the case of aqueous-phase processes, hydrothermal conditions result in transformation of  $\gamma$ - $Al_2O_3$  to boehmite, which is accompanied by degradation of the porous structure and leads to rapid catalyst deactivation.

of this work is to propose the simple one-stage approach to protect  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> porous structure from collapse under hydrothermal conditions.



Sample	C content, wt%	Graphitization degree, %
C@Al <sub>2</sub> O <sub>3</sub> -5	1.4	6
C@Al <sub>2</sub> O <sub>3</sub> -15	9.1	36
C@Al <sub>2</sub> O <sub>3</sub> -30	11.2	45
C@Al <sub>2</sub> O <sub>3</sub> -60	15.5	62
C@Al <sub>2</sub> O <sub>3</sub> -120	21.8	87
C@Al <sub>2</sub> O <sub>3</sub> -240	25.5	102

Treatment in C<sub>2</sub>H<sub>4</sub> flow at 680°C for 120 min and more leads to the formation of a monolayer carbon coating on the alumina surface.



Carbon in  $C@Al_2O_3$ samples, regardless of the graphitization time, is present in the form of nanocrystalline graphite with a graphene fragment size of less than 5 nm.



HRTEM data shows that the morphology of  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> consist of the mixture of lamellar



- $\succ$  An increase in graphitization time leads to a slight decrease in BET surface area of  $C@Al_2O_3$  samples.
- $\succ \gamma$ -Al<sub>2</sub>O<sub>3</sub> transforms into boehmite during hydrothermal treatment (XRD data).
- $\triangleright$  Hydrothermal treatment leads to a decrease in the initial  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> surface area by 80%.
- > A monolayer (or more) graphite coating of alumina under HTT contributes to the formation of a mesoporous C@boehmite composite with a high specific surface area similar to the initial material.



 $\blacktriangleright$  An increase in the duration of HTT from 5 to 72 h does not lead to additional structural

and textural changes, which indicates the stability of the formed C@boehmite

composite.

## Conclusion

Coating with carbon makes it possible to adapt  $\gamma$ -Al<sub>2</sub>O<sub>3</sub> to hydrothermal conditions, which makes  $C(a)Al_2O_3$  composites a promising support for catalysts of various aqueous-phase reactions.

M.A. Kazakova, A.G. Selyutin, M.V. Parfenov, A.V. Ishchenko, M.O. Kazakov, Micropor. Mesopor. Mat, 341 (2022) 112038.

The study was funded by Russian Science Foundation according to the research project No 21-73-10039, https://rscf.ru/project/21-73-10039/