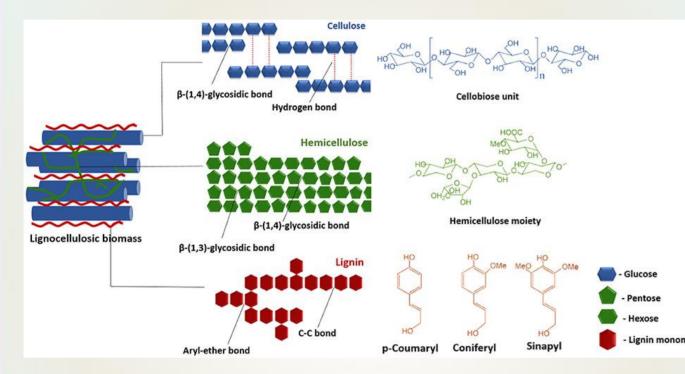
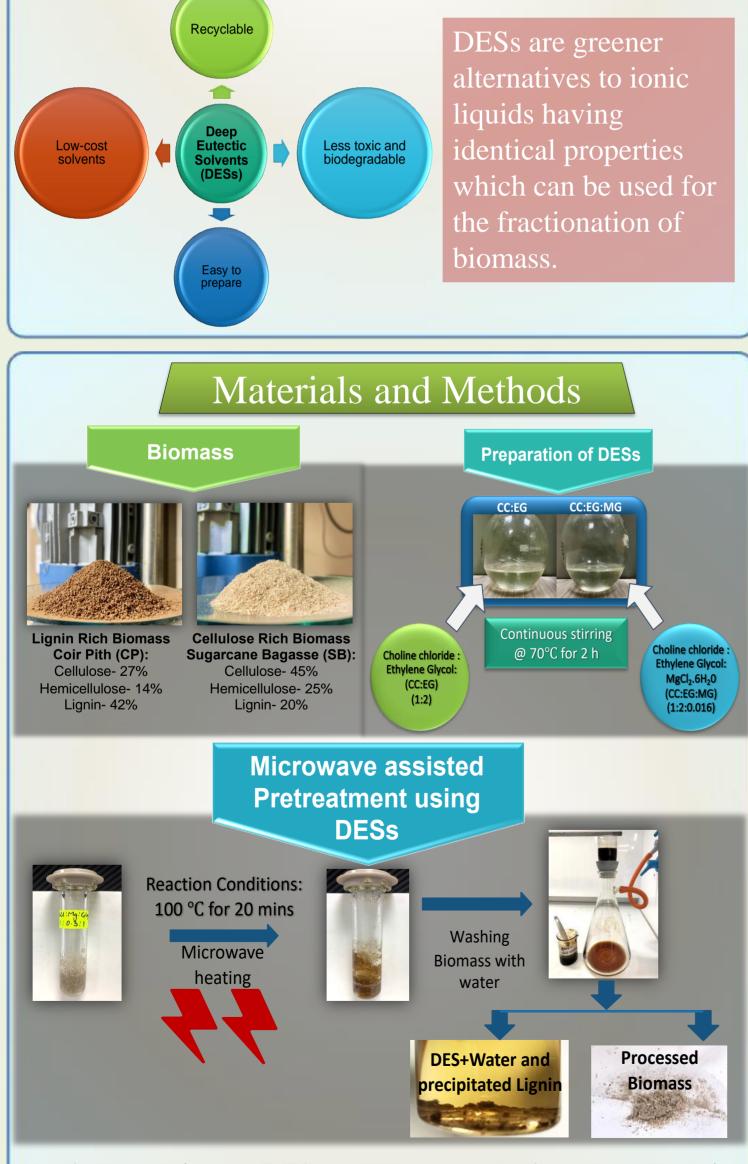
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Disintegration of lignocellulosic biomass using Deep eutectic solvents: INTERNATIONAL CONFERENCE on **Degradation kinetics and Py-GCMS study** BIOTECHNOLOGY FOR RESOURCE EFFICIENCY, ENERGY, ENVIRONMENT, CHEMICALS AND HEALTH Abhisek Sahoo^{1,*}, Thallada Bhaskar², Kamal K. Pant¹ ¹ - Department of Chemical Engineering, Indian Institute of Technology – Delhi ² - Thermo-catalytic Process Area, Indian Institute of Petroleum - Dehradun *sahooabhisek01@yahoo.com



- Lignocellulosic biomass (LBM) represents a renewable, widespread and low-cost source which can potentially be converted to fine chemicals and bio-fuels.
- Fractionation of biomass is the most energy intensive, expensive and challenging step of biomass valorisation.

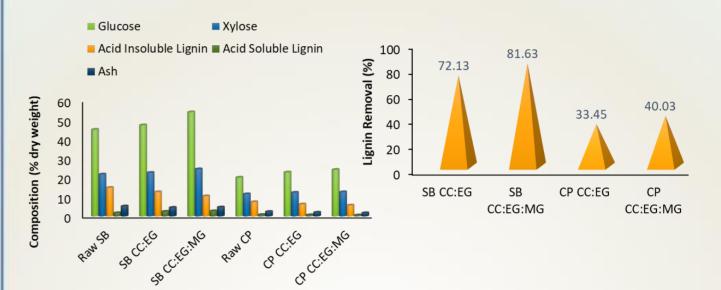




Results

• Physicochemical characterization:

SAMPLE	Volatile (wt.%)	Ash (wt.%)	C (%)	H (%)	S (%)	0 (%)	HHV (MJ/kg)
CP-RAW	70.42	7.86	52.74	3.83	0.24	41.75	18.44
CP-CE	71.74	1.4	55.00	4.59	0.19	37.61	20.67
CP-CEM	72.28	1.81	55.16	4.80	0.19	37.46	20.98
SB-RAW	80.60	7.24	45.56	3.99	0.26	48.80	15.42
SB-CE	82.53	6.32	50.96	4.59	0.19	42.69	18.65
SB-CEM	83.14	6.33	50.15	4.74	0.17	43.16	18.49



Degradation kinetics: NL-INT: $\left|\sum_{i=1}^{n}\sum_{j\neq i}^{n}\frac{\beta_{j}I(E_{\alpha},T_{\alpha,i})}{\beta_{i}I(E_{\alpha},T_{\alpha,i})}-n(n-1)\right|=\min(\Omega)$ **NL-DIF:** $\left| \sum_{i=1}^{n} \sum_{j \neq i}^{n} \frac{\beta_i \left(\frac{d\alpha}{dT}\right)_i \left[\exp\left(\frac{E_{\alpha}}{RT_{\alpha,i}}\right) \right]}{\beta_j \left(\frac{d\alpha}{dT}\right)_i \left[\exp\left(\frac{E_{\alpha}}{RT_{\alpha,i}}\right) \right]} - n(n-1) \right| = \min(\Omega)$ NL-INT NL-DIF 250 • NL-INT & NL-DIF

- •A three-constituent DES system was used to compare its efficiency with the popular binary DES (CC:EG).
- •Lewis acidic metal chloride salt (MgCl₂) was introduced to enhance the delignification process.

